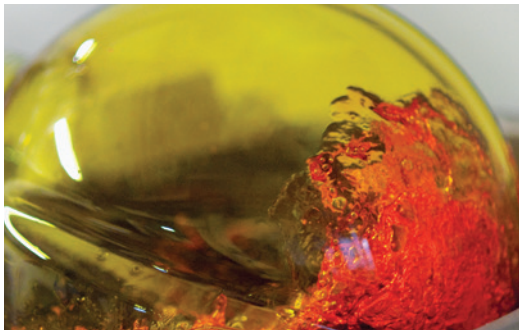
Precious Metal

Salts, Solutions and Anodes

METALOR®

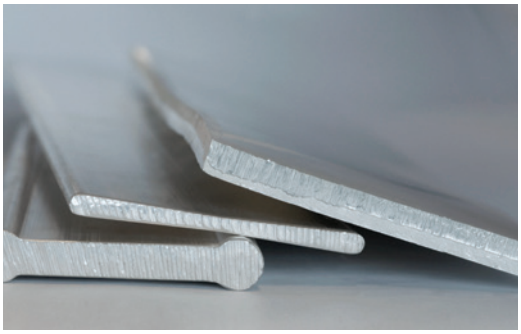
High-Quality Commodities for Superior Performance



Commercial Name	Metal	Formula	Chemical Name	CAS No.	PDS
Gold					**Au**
Gold grain Grenaille d'or	≥99.99 % ≥99.999 %	Au	gold	7440-57-5	856 866
Gold(I) cyanide* Cyanure d'or(I)	≥87.80 %	AuCN	gold(I) monocyanide	506-65-0	592
Gold(I) potassium cyanide (1) Cyanure d'or(I) et de potassium	≥68.20 %	$K[Au(CN)_2]$	potassium dicyanoaurate(I)	13967-50-5	591
Gold(I) potassium cyanide (2) (3) Cyanure d'or(I) et de potassium	≥68.00 %	$K[Au(CN)_2]$	potassium dicyanoaurate(I)	13967-50-5	980 1253
Gold(I) potassium cyanide (4) Cyanure d'or(I) et de potassium	≥68.00 %	$K[Au(CN)_2]$	potassium dicyanoaurate(I)	13967-50-5	741
Gold(I) potassium cyanide solution Cyanure d'or(I) et de potassium en solution	100 g/L	$K[Au(CN)_2]$	potassium dicyanoaurate(I)	13967-50-5	593
Gold(III) potassium cyanide Cyanure d'or(III) et de potassium	≥57.70 %	$K[Au(CN)_4]$	potassium tetracyanoaurate(III)	14263-59-3	597
Gold(III) potassium cyanide solution Cyanure d'or(III) et de potassium en solution	10 g/L	$K[Au(CN)_4]$	potassium tetracyanoaurate(III)	14263-59-3	833
Gold(III) potassium cyanide solution Cyanure d'or(III) et de potassium en solution	50 g/L	$K[Au(CN)_4]$	potassium tetracyanoaurate(III)	14263-59-3	832
Gold(III) potassium cyanide solution Cyanure d'or(III) et de potassium en solution	100 g/L	$K[Au(CN)_4]$	potassium tetracyanoaurate(III)	14263-59-3	680
Gold(I) ammonium sulphite solution Sulfite d'or(I) et d'ammonium en solution	100 g/L	NH_4AuSO_3 . n $(NH_4)_2SO_3$	ammonium gold(I) sulphite	71662-32-3	596
Gold(I) potassium sulphite solution Sulfite d'or(I) et de potassium en solution	100 g/L	$K_3[Au(SO_3)_2]$. n K_2SO_3	tripotassium gold(I) disulphite	19153-99-2	1093
Gold(I) sodium sulphite solution Sulfite d'or(I) et de sodium en solution	100 g/L	$Na_3[Au(SO_3)_2]$. n Na_2SO_3	trisodium gold(I) disulphite	19153-98-1	595
Gold(III) chloride hydrate* Chlorure d'or(III) hydraté	≥49.00 %	$H[AuCl_4]$. n H_2O	tetrachloroauric(III) acid	27988-77-8	594
Gold(III) chloride solution Chlorure d'or(III) en solution	100 g/L	$H[AuCl_4]$	tetrachloroauric(III) acid	27988-77-8	1104
Gold(III) chloride solution Chlorure d'or(III) en solution	200 g/L	$H[AuCl_4]$	tetrachloroauric(III) acid	27988-77-8	1105
Gold(III) chloride solution Chlorure d'or(III) en solution	250 g/L	$H[AuCl_4]$	tetrachloroauric(III) acid	27988-77-8	599
Gold(III) potassium chloride* Chlorure d'or(III) et de potassium	≥49.00 %	$K[AuCl_4]$	potassium tetrachloroaurate(III)	13682-61-6	598
Gold(III) sodium chloride hydrate* Chlorure d'or(III) et de sodium hydraté	49.00 - 49.50 %	$Na[AuCl_4]$. n H_2O	sodium tetrachloroaurate(III)	15189-51-2	712
Gold(III) sodium chloride solution Chlorure d'or(III) et de sodium en solution	200 g/kg	$Na[AuCl_4]$	sodium tetrachloroaurate(III)	15189-51-2	1815

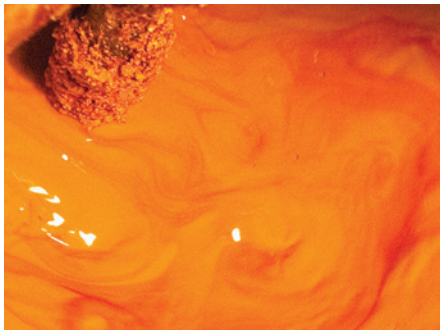
(1) for European market
(2) for Asian market
(3) for Japanese market
(4) for US market

* minimum order quantity

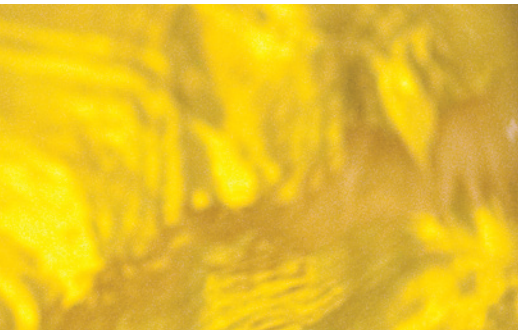
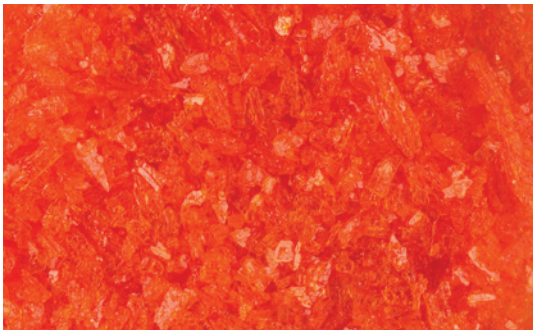


Commercial Name	Metal	Formula	Chemical Name	CAS No.	PDS
SILVER					**Ag**
Silver grain Grenaille d'argent	≥99.99 %	Ag	silver	7440-22-4	827
Silver crystals 140 - 5000 µm Cristaux d'argent 140 - 5000 µm	≥99.99 %	Ag	silver	7440-22-4	925
Silver anodes Solumax® Anodes d'argent	≥99.99 %	Ag	silver	7440-22-4	828
Silver sheet anodes Anodes d'argent	≥99.99 %	Ag	silver	7440-22-4	829
Silver(I) cyanide (1) Cyanure d'argent(I)	≥80.40 %	AgCN	silver(I) cyanide	506-64-9	582
Silver(I) cyanide (4) Cyanure d'argent(I)	≥80.00 %	AgCN	silver(I) cyanide	506-64-9	743
Silver(I) potassium cyanide (4) Cyanure d'argent(I) et de potassium	≥54.10 %	$K[Ag(CN)_2]$	potassium dicyanoargentate(I)	506-61-6	742
Silver(I) potassium cyanide (2) Cyanure d'argent(I) et de potassium	≥54.00 %	$K[Ag(CN)_2]$	potassium dicyanoargentate(I)	506-61-6	982
Silver(I) potassium cyanide (1) Cyanure d'argent(I) et de potassium	≥53.90 %	$K[Ag(CN)_2]$	potassium dicyanoargentate(I)	506-61-6	586
Silver(I) potassium cyanide (3) Cyanure d'argent(I) et de potassium	≥53.70 %	$K[Ag(CN)_2]$	potassium dicyanoargentate(I)	506-61-6	1254
Silver(I) sulphate Sulfate d'argent(I)	69.00 - 69.40 %	Ag_2SO_4	disilver(I) sulphate	10294-26-5	1251
Silver(I) sulphate ACS grade Sulfate d'argent(I)	68.80 - 69.30 %	Ag_2SO_4	disilver(I) sulphate	10294-26-5	721
Silver(I) chloride Chlorure d'argent(I)	≥75.20 %	AgCl	silver(I) chloride	7783-90-6	584
Silver(I) nitrate Nitrate d'argent(I)	≥63.40 %	$AgNO_3$	silver(I) nitrate	7761-88-8	834
Silver(I) oxide Oxyde d'argent(I)	≥92.80 %	Ag_2O	disilver(I) oxide	20667-12-3	587
Silver(I) acetate* Acétate d'argent(I)	≥63.00 %	$AgCH_3COO$	silver(I) acetate	563-63-3	588
Silver(I) carbonate* Carbonate d'argent(I)	≥77.90 %	Ag_2CO_3	disilver(I) carbonate	534-16-7	589
Silver(I) iodide* Iodure d'argent(I)	≥45.00 %	AgI	silver(I) iodide	7783-96-2	583



Commercial Name	Metal	Formula	Chemical Name	CAS No.	PDS
PALLADIUM					**Pd**
Palladium(II) chloride Chlorure de palladium(II)	≥59.30 %	$PdCl_2$	palladium(II) dichloride	7647-10-1	600
Palladium(II) chloride solution Chlorure de palladium(II) en solution	60 g/L	$PdCl_2$	palladium(II) dichloride	7647-10-1	839
Palladium(II) chloride solution Chlorure de palladium(II) en solution	100 g/L	$PdCl_2$	palladium(II) dichloride	7647-10-1	607
Palladium(II) chloride solution Chlorure de palladium(II) en solution	200 g/L	$PdCl_2$	palladium(II) dichloride	7647-10-1	840
Palladium(II) dichlorodiammine Dichlorodiammine de palladium(II)	≥49.80 %	$Pd(NH_3)_2Cl_2$	diamminepalladium(II) dichloride	14323-43-4	604
Palladium(II) dichlorotetrammine Dichlorotétrammine de palladium(II)	≥40.50 %	$Pd(NH_3)_4Cl_2$	tetraamminepalladium(II) dichloride	13815-17-3	605
Palladium(II) dichlorotetrammine solution Dichlorotétrammine de palladium(II) en solution	100 g/L	$Pd(NH_3)_4Cl_2$	tetraamminepalladium(II) dichloride	13815-17-3	608
Palladium(II) dinitrodiammine Dinitrodiammine de palladium(II)	≥45.00 %	$Pd(NH_3)_2(NO_2)_2$	diamminepalladium(II) dinitrite	14852-83-6	606
Palladium(II) dinitrodiammine solution Dinitrodiammine de palladium(II) en solution	150 g/L	$Pd(NH_3)_2(NO_2)_2$	diamminepalladium(II) dinitrite	14852-83-6	993
Palladium(II) tetrammine sulphate Sulfate de palladium(II) tétrammine	≥39.00 %	$Pd(NH_3)_4SO_4$	tetraamminepalladium(II) sulphate	13601-06-4	602
Palladium(II) tetrammine sulphate solution Sulfate de palladium(II) tétrammine en solution	40 g/L	$Pd(NH_3)_4SO_4$	tetraamminepalladium(II) sulphate	13601-06-4	1279
Palladium(II) tetrammine sulphate solution Sulfate de palladium(II) tétrammine en solution	50 g/L	$Pd(NH_3)_4SO_4$	tetraamminepalladium(II) sulphate	13601-06-4	1277
Palladium(II) nitrate solution Nitrate de palladium(II) en solution	100 g/L	$Pd(NO_3)_2$	palladium(II) dinitrate	10102-05-3	761
Palladium(II) nitrate solution Nitrate de palladium(II) en solution	200 g/L	$Pd(NO_3)_2$	palladium(II) dinitrate	10102-05-3	773
Palladium(II) sulphate solution Sulfate de palladium(II) en solution	100 g/L	$PdSO_4$	palladium(II) sulphate	13566-03-5	855
Palladium(II) oxide hydrate Oxyde de palladium(II)	67.50 - 74.50 %	$PdO \cdot H_2O$	palladium(II) oxide	64109-12-2	844
Palladium(II) acetate Acétate de palladium(II)	≥46.60 %	$Pd(CH_3COO)_2$	palladium(II) acetate	3375-31-3	1858
Potassium tetrachloropalladate(II)* Tétrachloropalladate(II) de potassium	≥30.00 %	$K_2[PdCl_4]$	dipotassium tetrachloropalladate(II)	10025-98-6	907
Sodium tetrachloropalladate(II)* Tétrachloropalladate(II) de sodium	≥34.00 %	$Na_2[PdCl_4]$	disodium tetrachloropalladate(II)	13820-53-6	842
Sodium tetrachloropalladate(II) solution Tétrachloropalladate(I) de sodium en solution	120 g/kg	$Na_2[PdCl_4]$	disodium tetrachloropalladate(II)	13820-53-6	1900


Commercial Name	Metal	Formula	Chemical Name	CAS No.	PDS
PLATINUM					**Pt**
Ammonium hexachloroplatinate(IV) Hexachloroplatinate(IV) d'ammonium	≥42.00 %	$(NH_4)_2[PtCl_6]$	diammonium hexachloroplatinate(IV)	16919-58-7	780
Chloroplatinic(IV) acid hydrate* Acide chloroplatinique(IV) hydraté	≥37.40 %	$H_2[PtCl_6] \cdot n\ H_2O$	hexachloroplatinic(IV) acid	16941-12-1	609 1295
Chloroplatinic(IV) acid solution Acide chloroplatinique(IV) en solution	50 g/L	$H_2[PtCl_6]$	hexachloroplatinic(IV) acid	16941-12-1	824
Chloroplatinic(IV) acid solution Acide chloroplatinique(IV) en solution	100 g/L	$H_2[PtCl_6]$	hexachloroplatinic(IV) acid	16941-12-1	949
Hexahydroxyplatinic(IV) acid Acide hexahydroxyplatinique(IV)	≥60.20 %	$H_2[Pt(OH)_6]$	dihydrogen hexahydroxyplatinate(IV)	51850-20-5	581
Platinum "P salt" solution "Sel P" de platine en solution	25 g/L	$Pt(NH_3)_2(NO_2)_2$	diammineplatinum(II) dinitrite	14286-02-3	729
Sodium hexachloroplatinate(IV) hexahydrate* Chloroplatinate(IV) de sodium hexahydraté	≥34.00 %	$Na_2[PtCl_6] \cdot 6H_2O$	disodium hexachloroplatinate(IV)	19583-77-8	611
Potassium hexachloroplatinate(IV)* Hexachloroplatinate(IV) de potassium	≥39.80 %	$K_2[PtCl_6]$	dipotassium hexachloroplatinate(IV)	16921-30-5	779
RHODIUM					**Rh**
Rhodium(III) chloride trihydrate* Chlorure de rhodium(III) trihydraté	≥39.00 %	$RhCl_3 \cdot 3H_2O$	rhodium(III) trichloride	13569-65-8	618
Rhodium(III) iodide* Iodure de rhodium(III)	≥18.50 %	RhI_3	rhodium(III) triiodide	15492-38-3	697
Rhodium(III) sulphate solution Sulfate de rhodium(III) en solution	50 g/L	$Rh_2(SO_4)_3$	dirhodium(III) trisulphate	10489-46-0	914
Rhodium(III) nitrate solution Nitrate de rhodium(III) en solution	50 g/L	$Rh(NO_3)_2$	rhodium(III) trinitrate	10139-58-9	1912
IRIDIUM					**Ir**
Hexachloroiridic(IV) acid hydrate* Acide hexachloroiridique(IV) hydraté	≥35.00 %	$H_2[IrCl_6] \cdot n\ H_2O$	hexachloroiridic(IV) acid	16941-92-7	693
Iridium(III) chloride hydrate* Chlorure d'iridium(III) hydraté	≥47.00 %	$IrCl_3 \cdot n\ H_2O$	iridium(III) trichloride	14996-61-3	723
RUTHENIUM					**Ru**
Ruthenium(III) chloride hydrate Chlorure de ruthénium(III) hydraté	39.0 - 42.0 %	$RuCl_3 \cdot n\ H_2O$	ruthenium(III) trichloride	14898-67-0	619

* minimum order quantity

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